



## LJMU Research Online

**Wallach, J, Kavanagh, PV, Mclaughlin, G, Morris, N, Power, JD, Elliott, SP, Mercier, MS, Lodge, D, Morris, H, Dempster, NM and Brandt, SD**

**Preparation and characterization of the 'research chemical' diphenidine, its pyrrolidine analogue, and their 2,2-diphenylethyl isomers**

<http://researchonline.ljmu.ac.uk/id/eprint/3408/>

### Article

**Citation** (please note it is advisable to refer to the publisher's version if you intend to cite from this work)

**Wallach, J, Kavanagh, PV, Mclaughlin, G, Morris, N, Power, JD, Elliott, SP, Mercier, MS, Lodge, D, Morris, H, Dempster, NM and Brandt, SD (2015) Preparation and characterization of the 'research chemical' diphenidine, its pyrrolidine analogue, and their 2,2-diphenylethyl isomers. Drug Testina and**

LJMU has developed [LJMU Research Online](http://researchonline.ljmu.ac.uk/) for users to access the research output of the University more effectively. Copyright © and Moral Rights for the papers on this site are retained by the individual authors and/or other copyright owners. Users may download and/or print one copy of any article(s) in LJMU Research Online to facilitate their private study or for non-commercial research. You may not engage in further distribution of the material or use it for any profit-making activities or any commercial gain.

The version presented here may differ from the published version or from the version of the record. Please see the repository URL above for details on accessing the published version and note that access may require a subscription.

For more information please contact [researchonline@ljmu.ac.uk](mailto:researchonline@ljmu.ac.uk)

<http://researchonline.ljmu.ac.uk/>

## Preparation and characterization of the 'research chemical' diphenidine, its pyrrolidine analogue, and their 2,2-diphenylethyl isomers

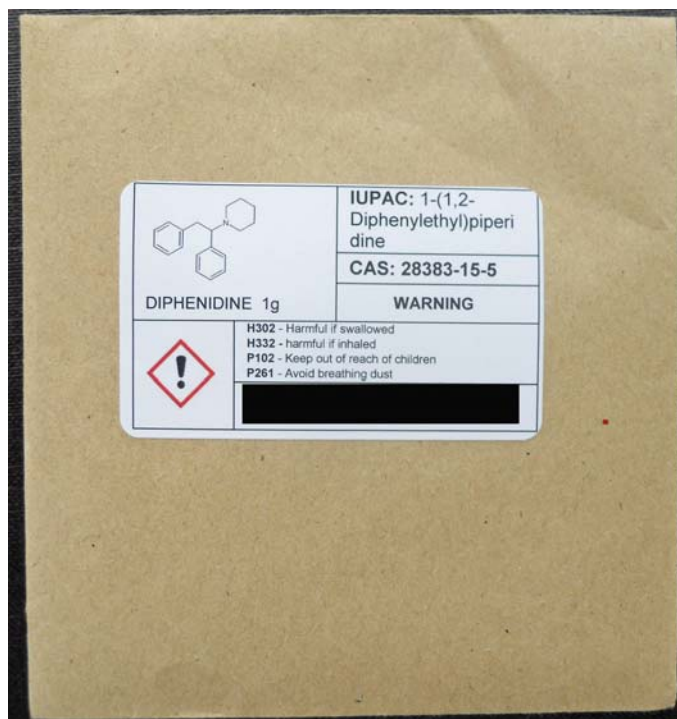
Jason Wallach, Pierce V. Kavanagh, Gavin McLaughlin, Noreen Morris, John D. Power, Simon P. Elliott, Marion S. Mercier, David Lodge, Hamilton Morris, Nicola M. Dempster, Simon D. Brandt

*Correspondence to: Simon D. Brandt, School of Pharmacy and Biomolecular Sciences, Liverpool John Moores University, Byrom Street, Liverpool, L3 3AF, UK. E-Mail: s.brandt@ljmu.ac.uk*

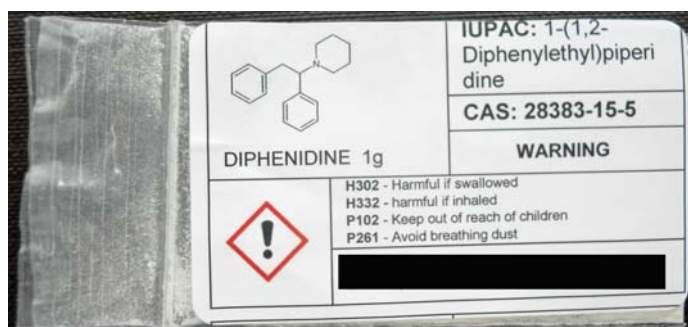
### Contents

- 1) Representative photograph of a diphenidine product obtained online
- 2) ATR-IR and NMR data of 1,2-diphenylethanamine HCl & 2,2-diphenylethanamine HCl
- 3) ATR-IR of 1-(1,2-diphenylethyl)piperidine (1,2-DEP, diphenidine) **(1)** and 1-(2,2-diphenylethyl)piperidine (2,2-DEP, diphenidine isomer) **(2)**
- 4) ATR-IR of 1-(1,2-diphenylethyl)pyrrolidine (1,2-DEPy) **(3)** and 1-(2,2-diphenylethyl)pyrrolidine (2,2-DEPy) **(4)**
- 5) ATR-IR of two diphenidine samples obtained online
- 6) GC-MS and EI ion trap MS spectra of 1,2-diphenylethanamine HCl & 2,2-diphenylethanamine HCl
- 7) ESI-triple quadrupole tandem mass spectra of 1,2-diphenylethanamine HCl & 2,2-diphenylethanamine HCl
- 8) LC-ESI-MS/MS chromatograms of 1,2-diphenylethanamine HCl & 2,2-diphenylethanamine HCl
- 9) LC-ESI-MS/MS chromatograms 1-(1,2-diphenylethyl)pyrrolidine (1,2-DEPy) **(3)** and 1-(2,2-diphenylethyl)pyrrolidine (2,2-DEPy) **(4)**
- 10) GC-EI-MS traces of **(1)** – **(4)** and two diphenidine samples obtained online
- 11) LC-ESI-MS/MS traces of synthesized **(1)** and **(2)** and two diphenidine samples obtained online
- 12) ESI-triple quadrupole tandem mass spectra of two diphenidine samples obtained online
- 13)  $^1\text{H}$  and  $^{13}\text{C}$  NMR of diphenidine free base and hydrochloride salt
- 14) Assigned  $^{13}\text{C}$  and  $^1\text{H}$  aliphatic and chemical shifts in ppm for compounds **(1)** – **(4)** and **(1)** HCl
- 15) HPLC-DAD traces for synthesized diphenidine isomers **(1)** and **(2)** and comparison with a diphenidine sample obtained from an online test purchase

1) Representative photograph of a diphenidine product obtained online

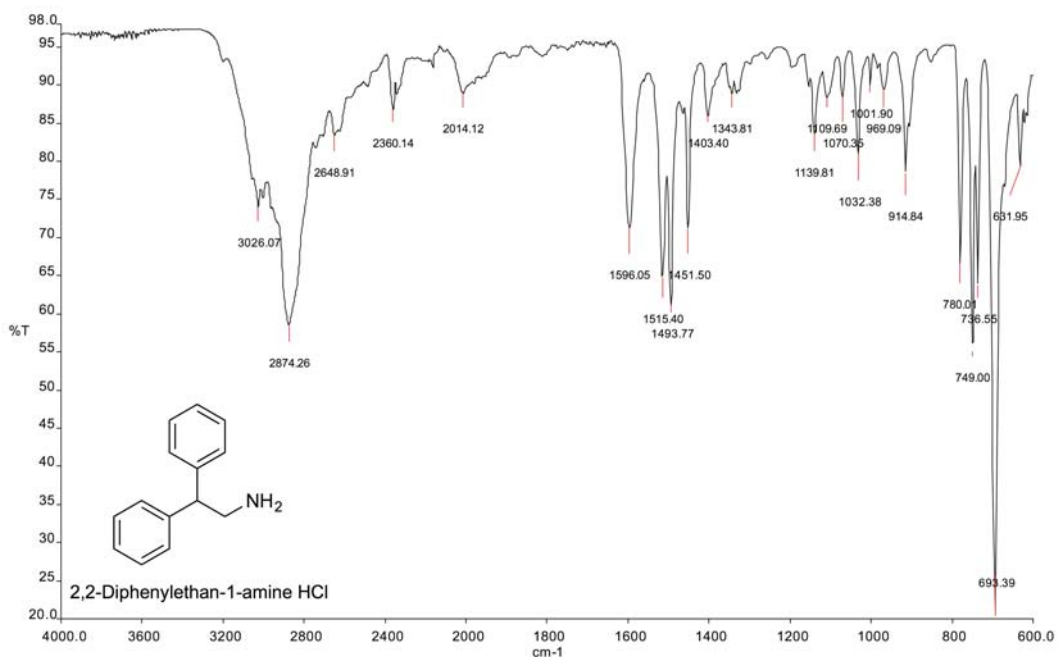
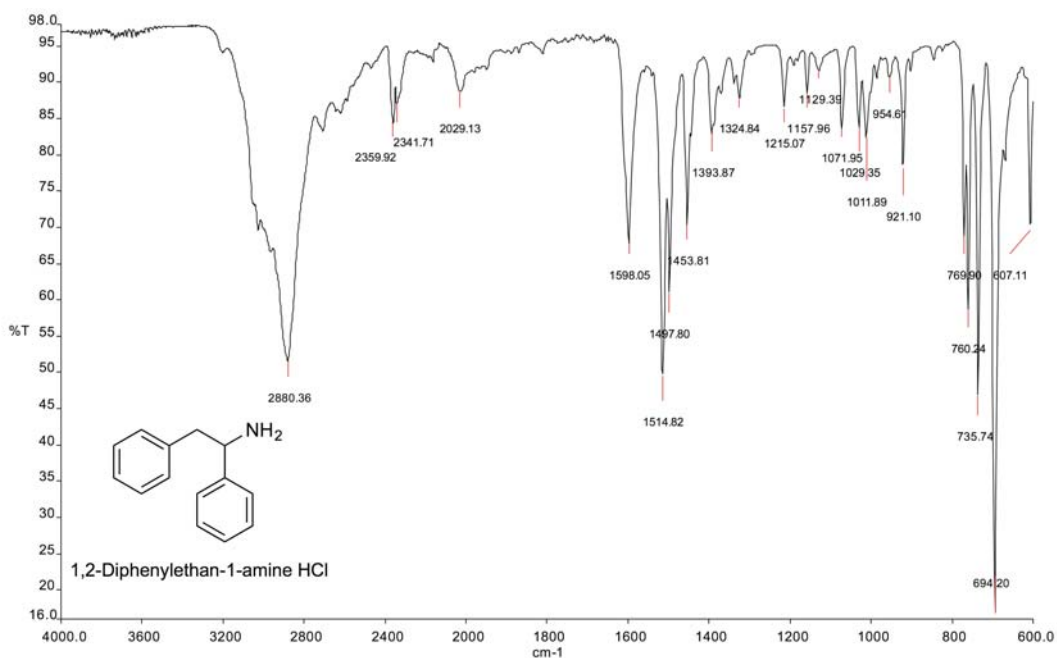


Note: CAS number refers to the hydrochloride salt of diphenidine



Infrared (IR) spectra were obtained on a Perkin Elmer Spectrum BX FTIR model using a Pike MIRacle ATR system. Data were acquired with the Spectrum v5.01 software (scan range 4000\_400  $\text{cm}^{-1}$ , resolution 4  $\text{cm}^{-1}$ , 16 scans).

## 2) ATR-IR and NMR data of 1,2-diphenylethanamine HCl &amp; 2,2-diphenylethanamine HCl



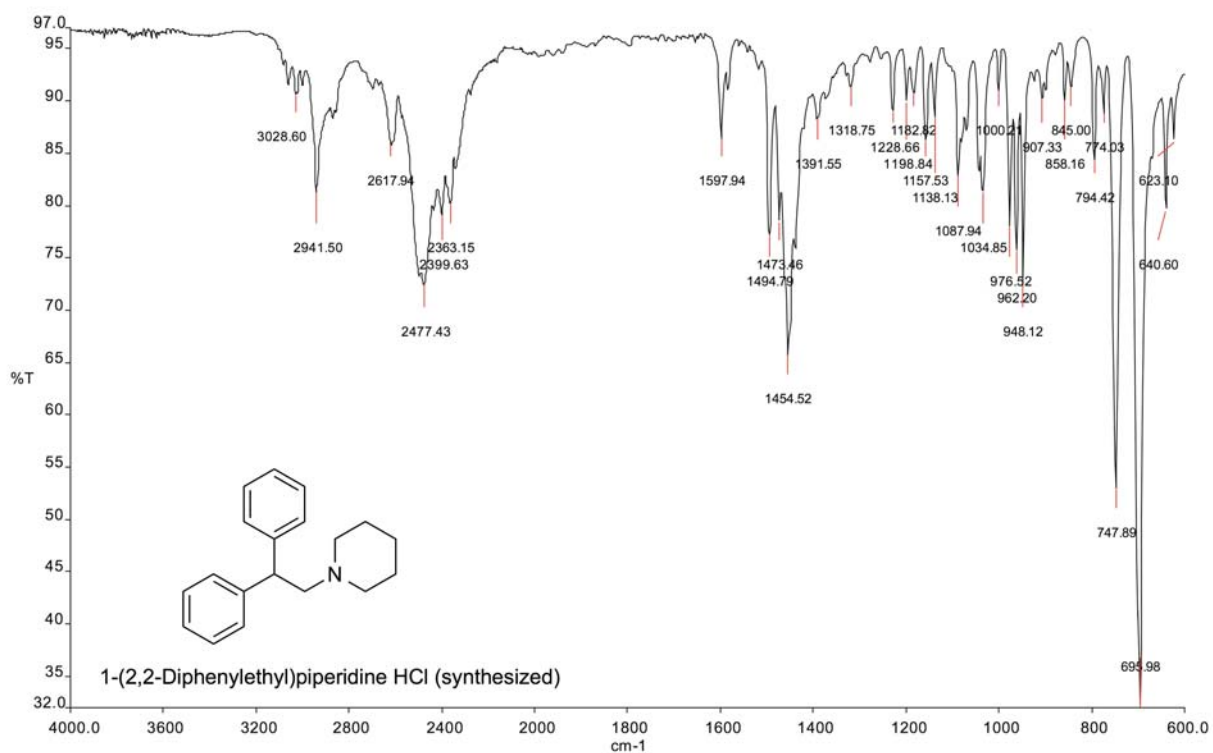
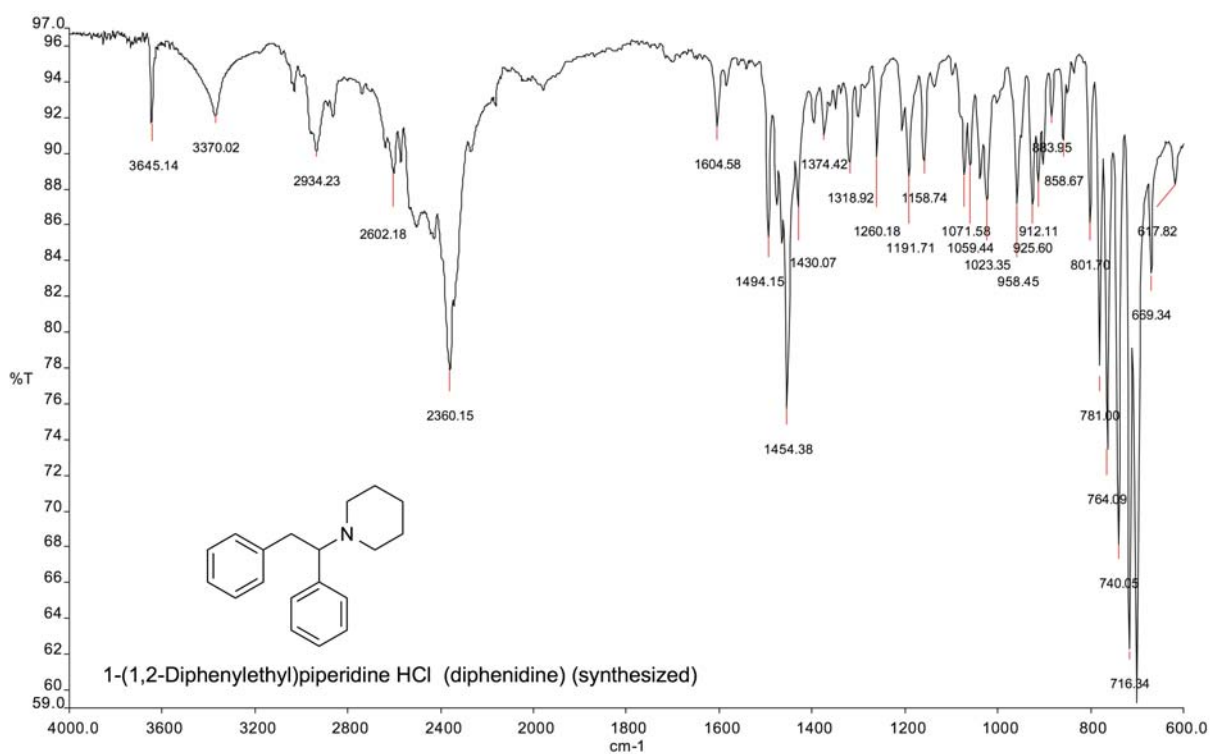
NMR 1,2-diphenylethanamine base (m.p. HCl salt 252.5-254.0 °C):

$^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.40-7.11 (10H, m, Ar-H), 4.19 (1H, dd,  $J = 8.8, 4.9$  Hz,  $\text{C}_1\text{H}$ ), 3.01 (1H, dd,  $J = 13.3, 4.9$  Hz,  $\text{C}_2\text{H}$ ), 2.83 (1H, dd,  $J = 13.3, 8.8$  Hz,  $\text{C}_2\text{H}$ ), 1.4 (2H, s,  $\text{NH}_2$ ).  
 $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  145.64 (quat. Ar-C), 139.09 (quat. Ar-C), 129.34 (Ar-CH), 128.39 (Ar-CH), 127.04 (Ar-CH), 126.42 (Ar-CH), 126.35 (Ar-CH), 57.55 ( $\text{C}_1\text{H}$ ), 46.59 ( $\text{CH}_2$ ).

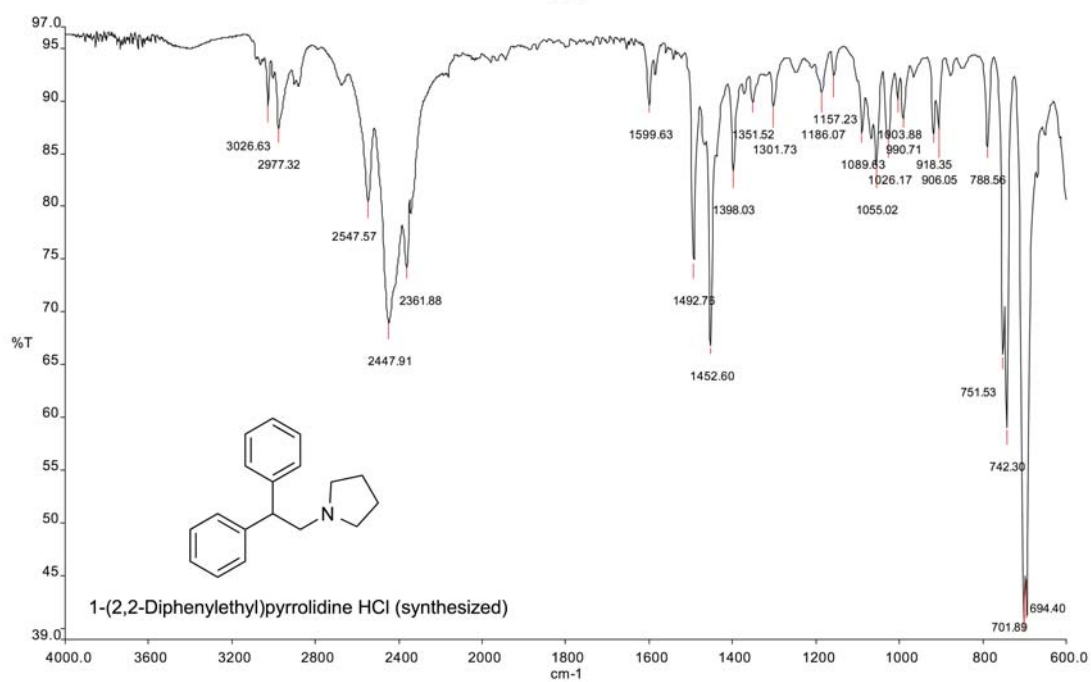
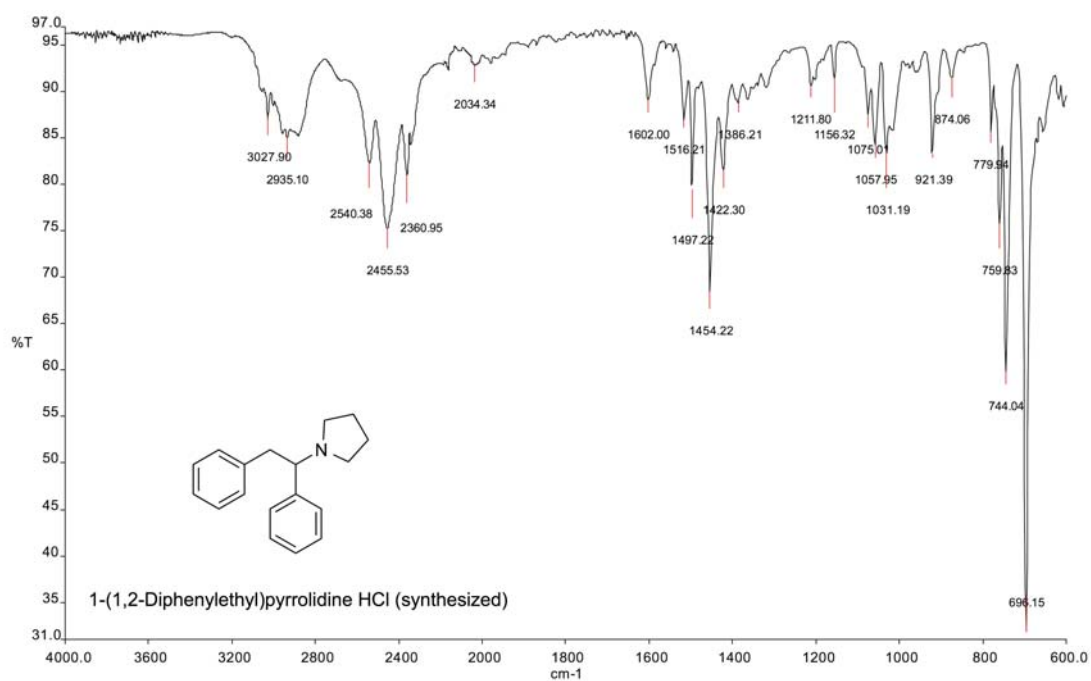
NMR 2,2-diphenylethanamine base (m.p. HCl salt > 260 °C):

$^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.34-7.14 (10H, m, Ar-H), 3.98 (1H, t,  $J = 7.6$  Hz,  $\text{C}_2\text{H}$ ), 3.31 (2H, d,  $J = 7.6$  Hz,  $\text{C}_1\text{H}_2$ ), 1.45 (2H, s,  $\text{NH}_2$ ).  
 $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  142.69 (2 x quat. Ar-C), 128.61 (4 x Ar-CH), 128.08 (4 x Ar-CH), 126.52 (2 x Ar-CH), 55.02 ( $\text{C}_1\text{H}_2$ ), 47.01 ( $\text{C}_2\text{H}$ ).

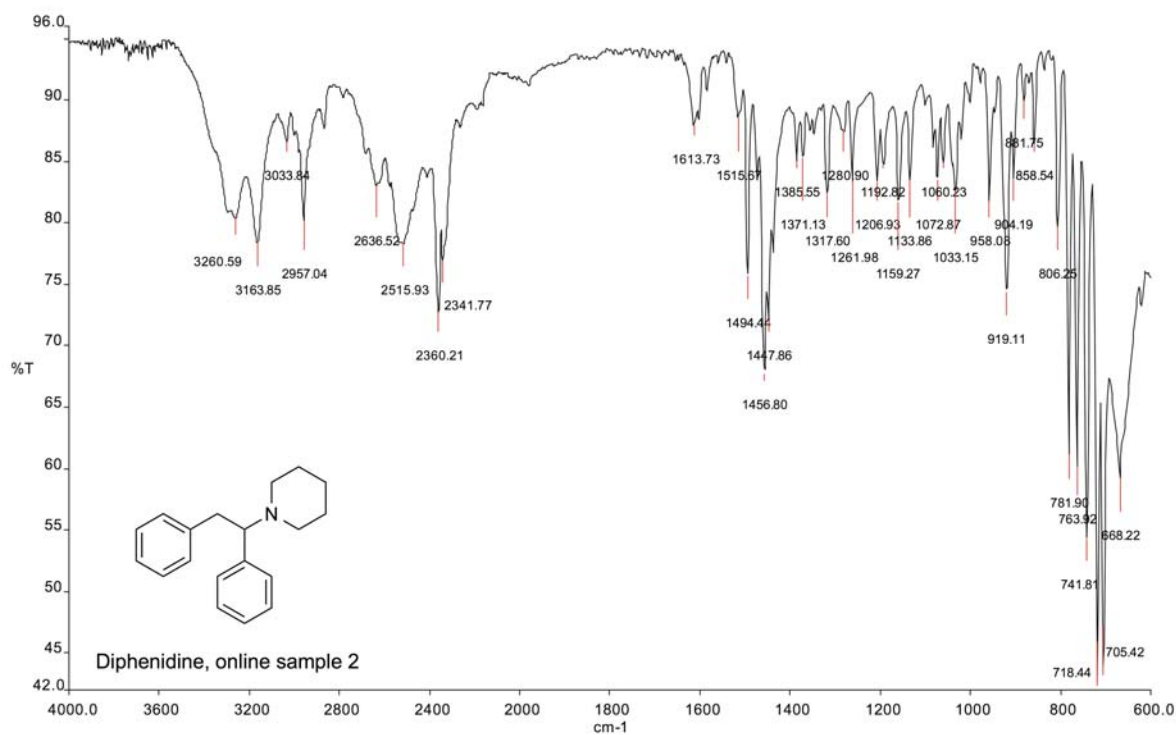
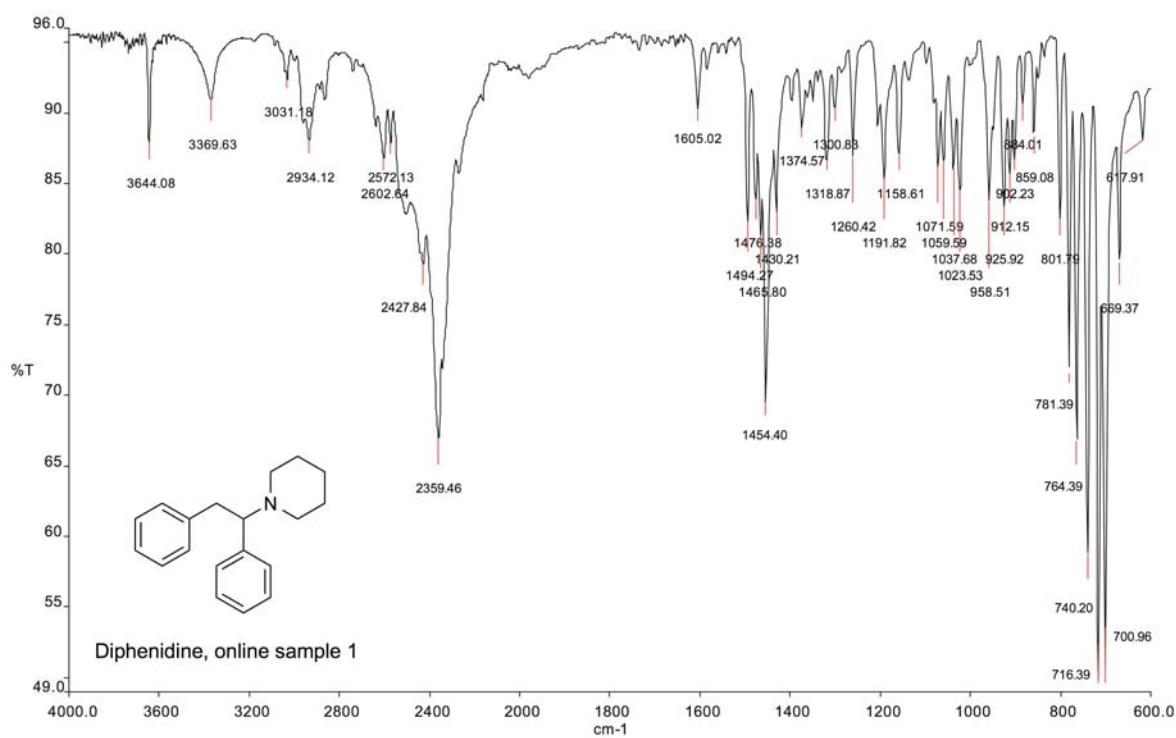
3) ATR-IR of 1-(1,2-diphenylethyl)piperidine (1,2-DEP, diphenidine) (1) and 1-(2,2-diphenylethyl)piperidine (2,2-DEP, diphenidine isomer) (2)



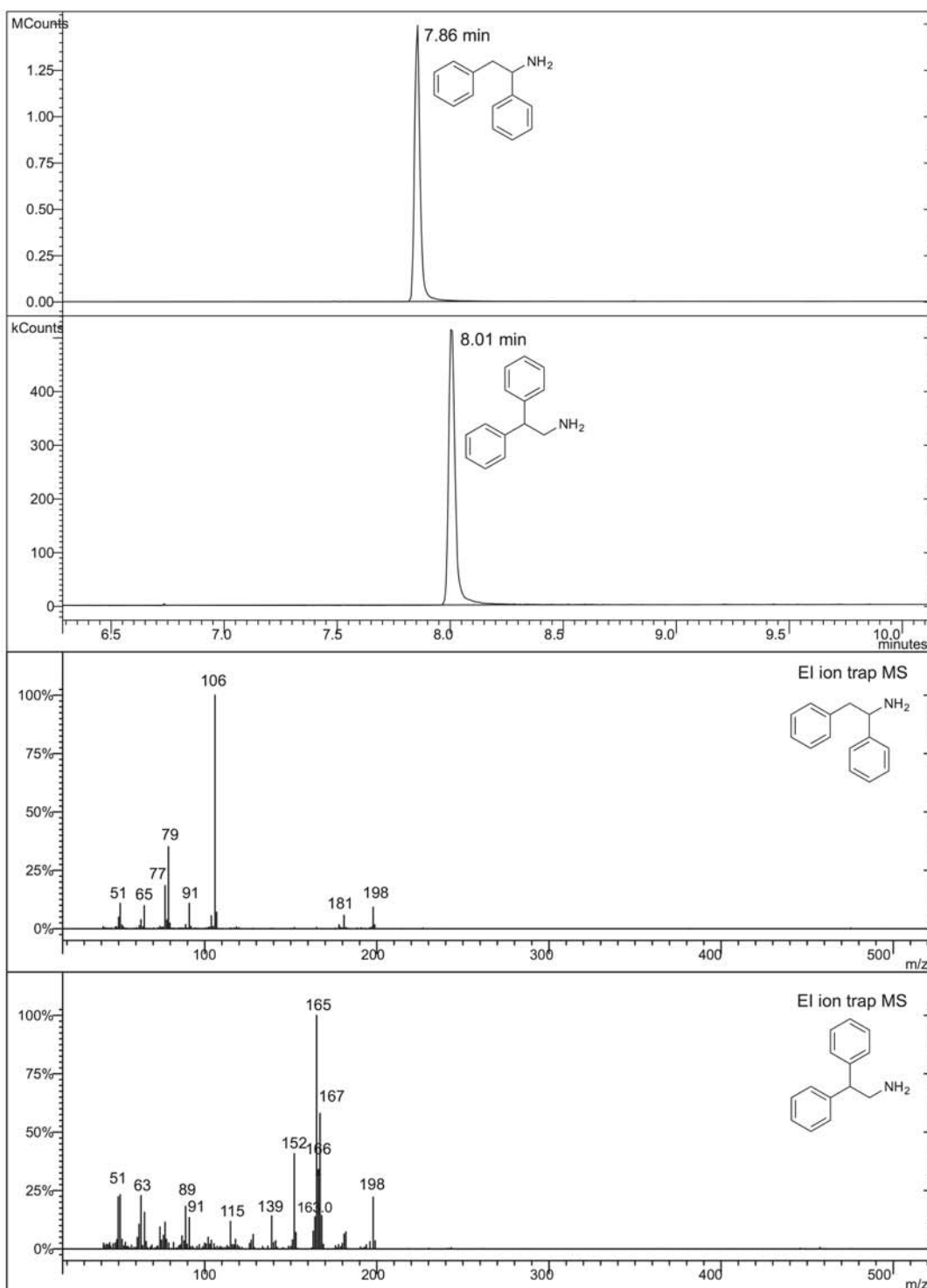
4) ATR-IR of 1-(1,2-diphenylethyl)pyrrolidine (1,2-DEPy) (**3**) and 1-(2,2-diphenylethyl)pyrrolidine (2,2-DEPy) (**4**)



5) ATR-IR of two diphenidine samples obtained online



6) GC-MS and EI ion trap MS spectra of 1,2-diphenylethanamine HCl & 2,2-diphenylethanamine HCl

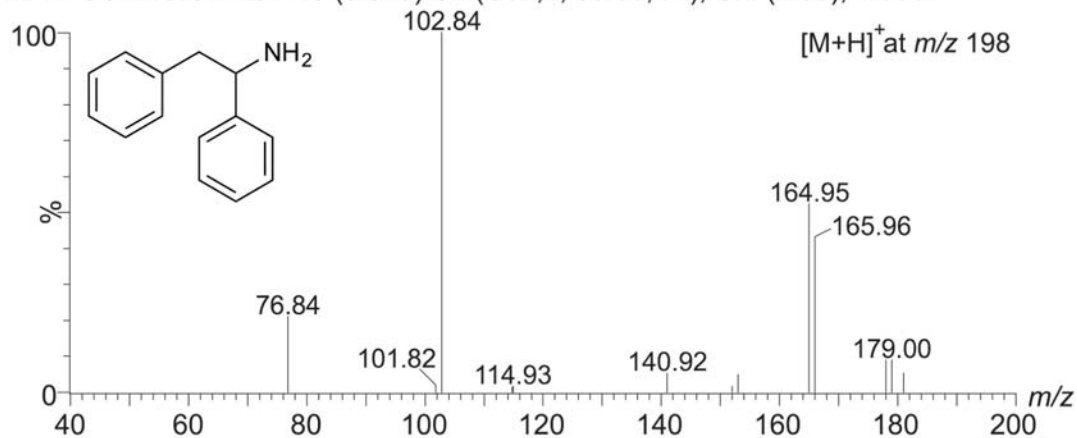




7) ESI-triple quadrupole tandem mass spectra of 1,2-diphenylethanamine HCl & 2,2-diphenylethanamine HCl

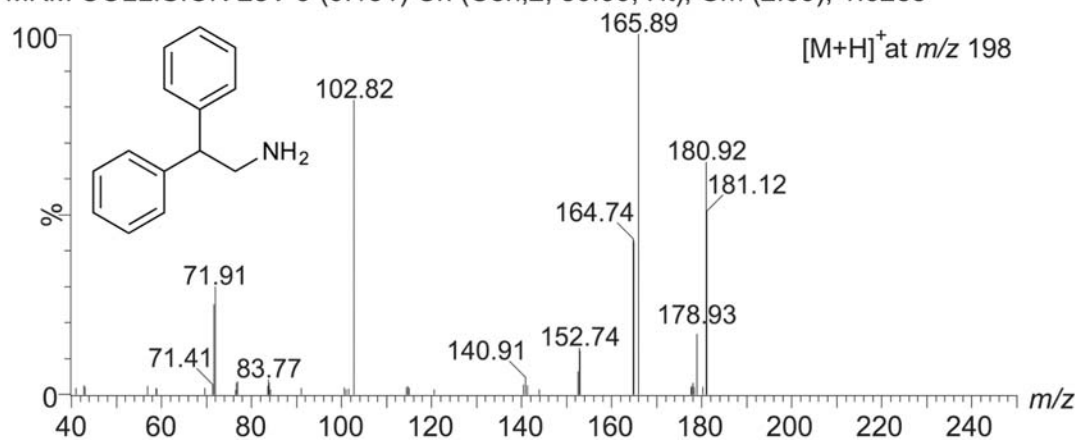
Cone 14V

MRM COLLISION 28V 49 (0.825) Cn (Cen,2, 80.00, Ht); Cm (2:59), 1.05e7

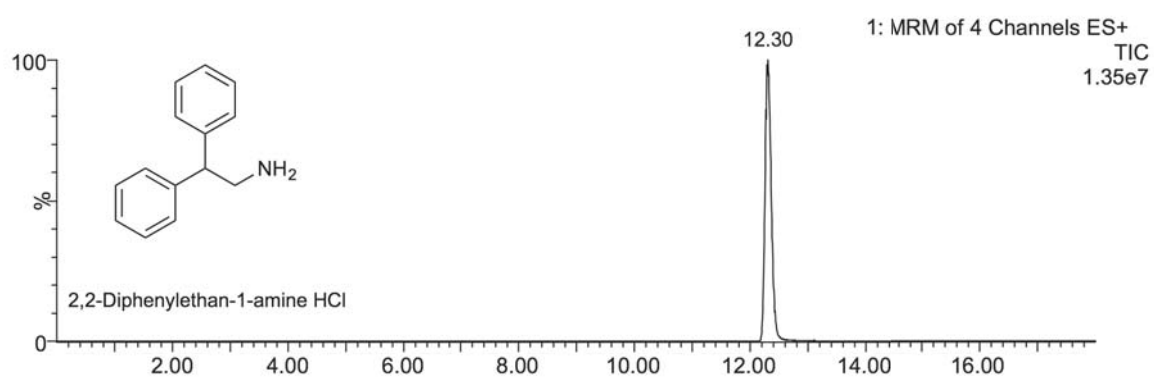
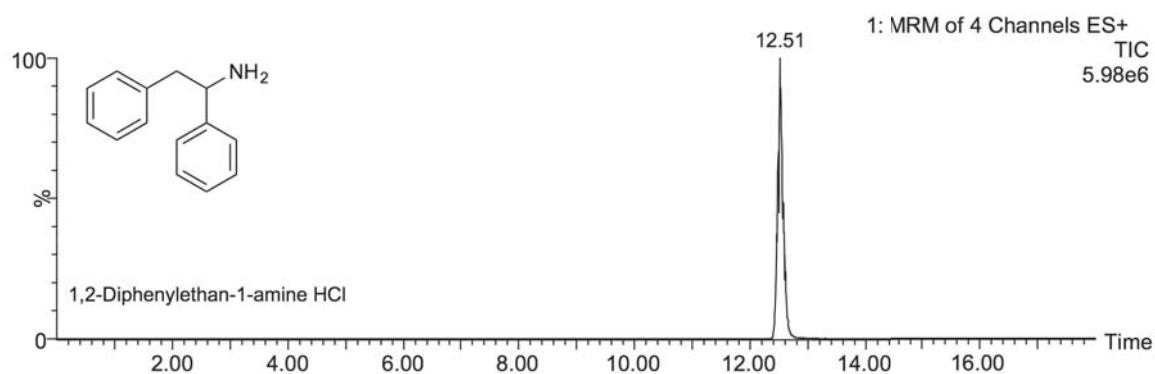


Cone 16V

MRM COLLISION 28V 9 (0.151) Cn (Cen,2, 80.00, Ht); Cm (2:60), 1.62e5



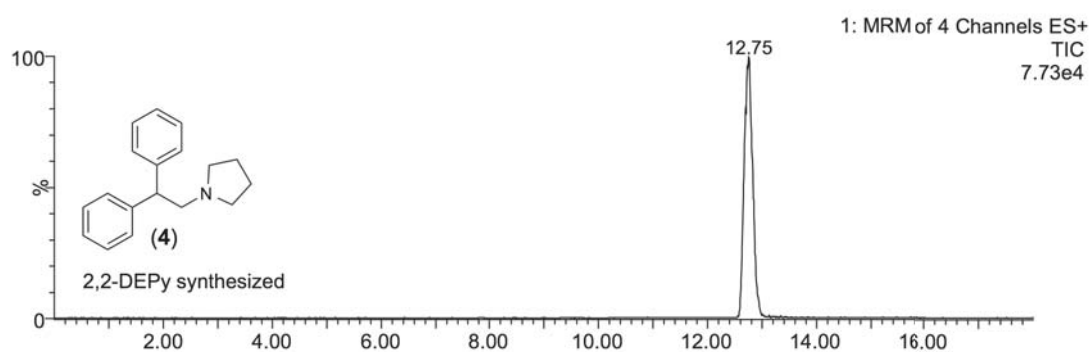
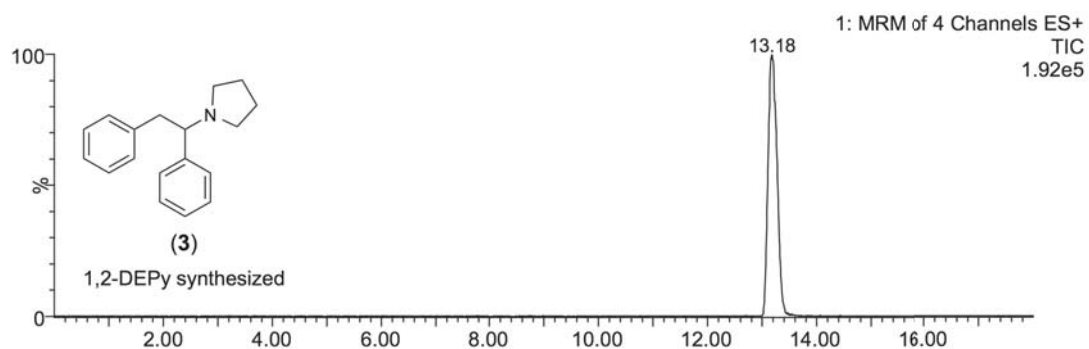
8) LC-ESI-MS/MS chromatograms of 1,2-diphenylethanamine HCl & 2,2-diphenylethanamine HCl



Selected ion transitions:

- $m/z$  198 > 72 (48 eV)
- $m/z$  198 > 103 (35 eV)
- $m/z$  198 > 166 (28 eV)
- $m/z$  198 > 181 (20 eV)

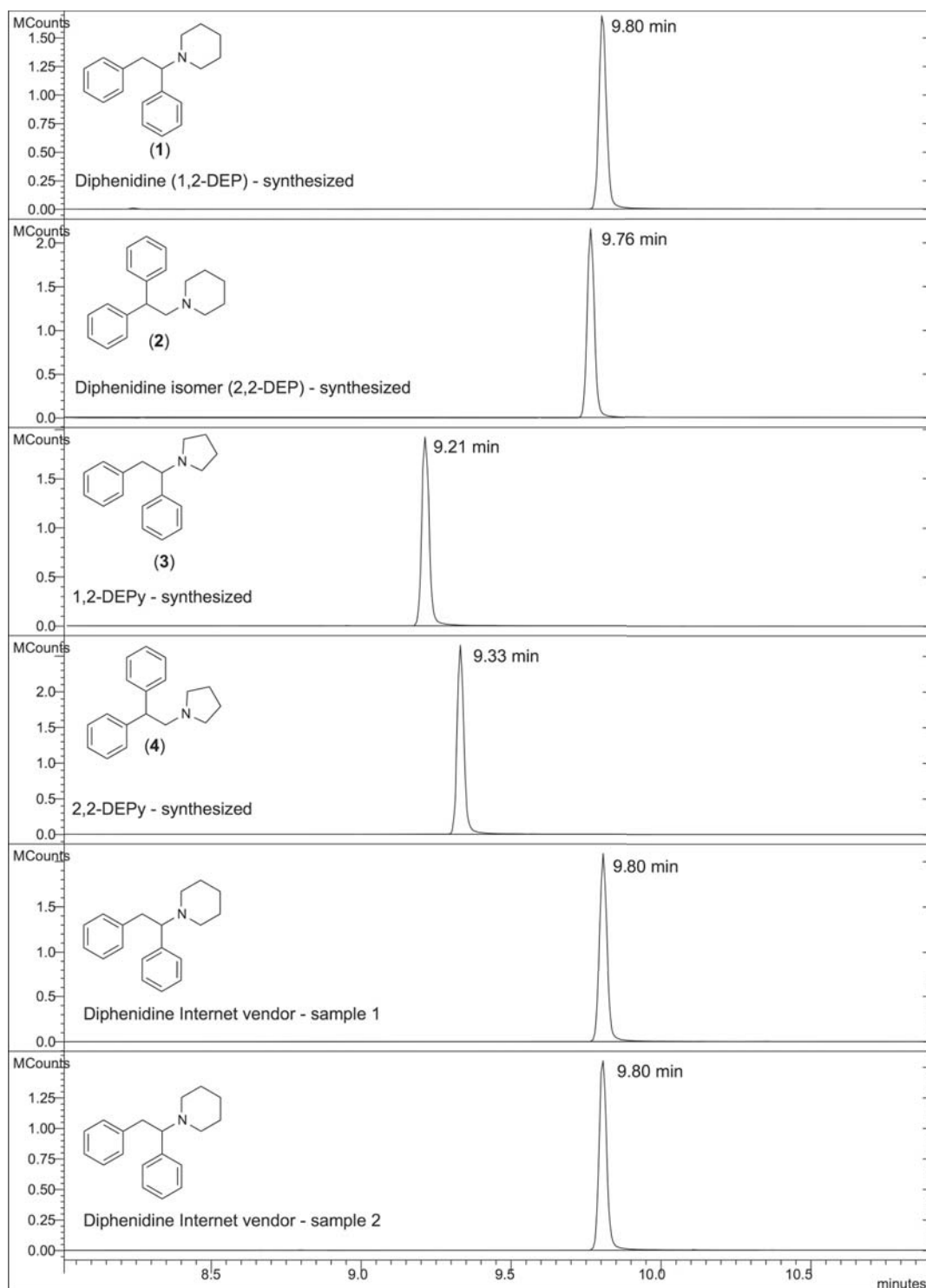
9) LC-ESI-MS/MS chromatograms 1-(1,2-diphenylethyl)pyrrolidine (1,2-DEPy) (**3**) and 1-(2,2-diphenylethyl)pyrrolidine (2,2-DEPy) (**4**)



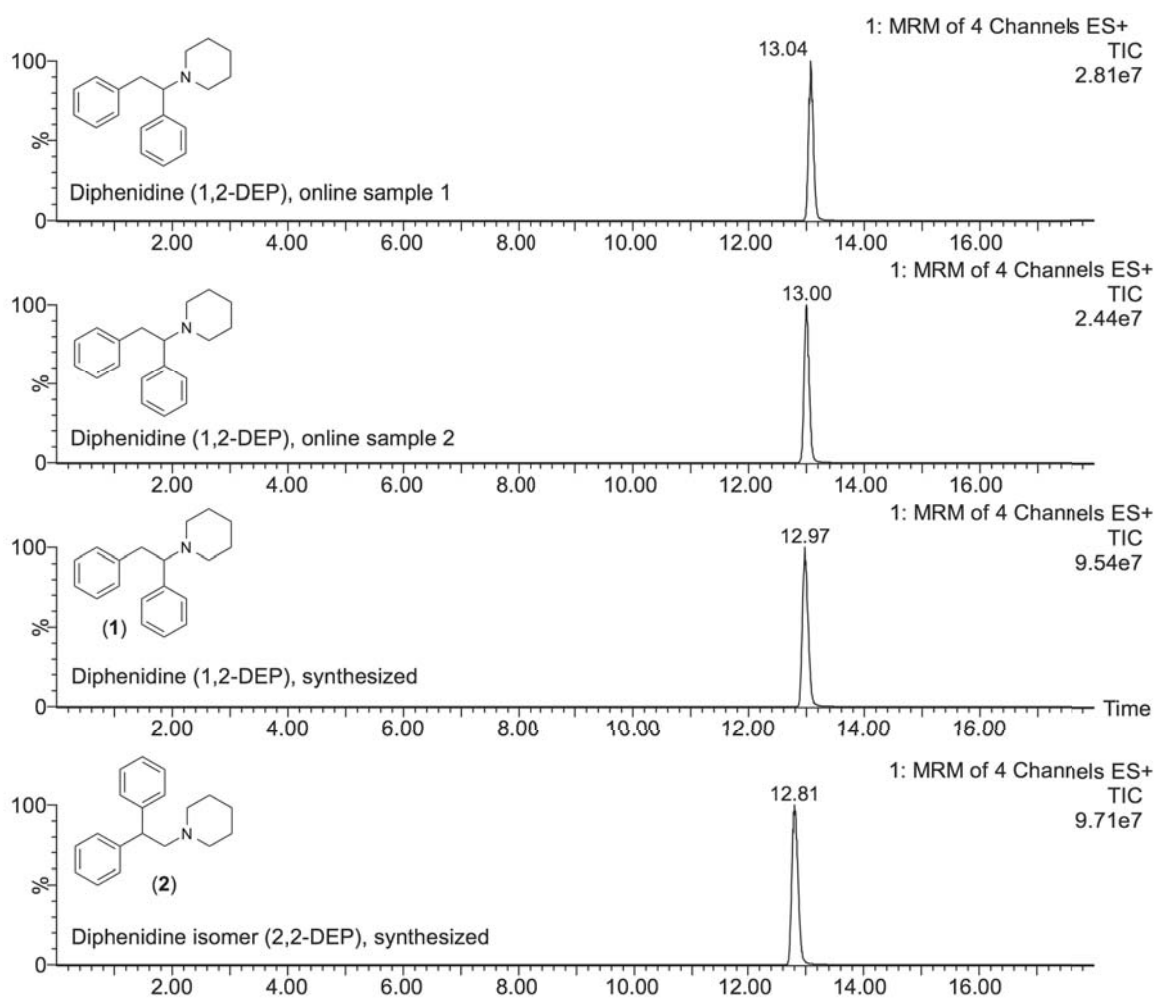
Selected ion transitions:

- $m/z$  252 > 72 (48 eV)
- $m/z$  252 > 103 (35 eV)
- $m/z$  252 > 166 (28 eV)
- $m/z$  252 > 181 (20 eV)

10) GC-EI-MS traces of (1) – (4) and two diphenidine samples obtained online



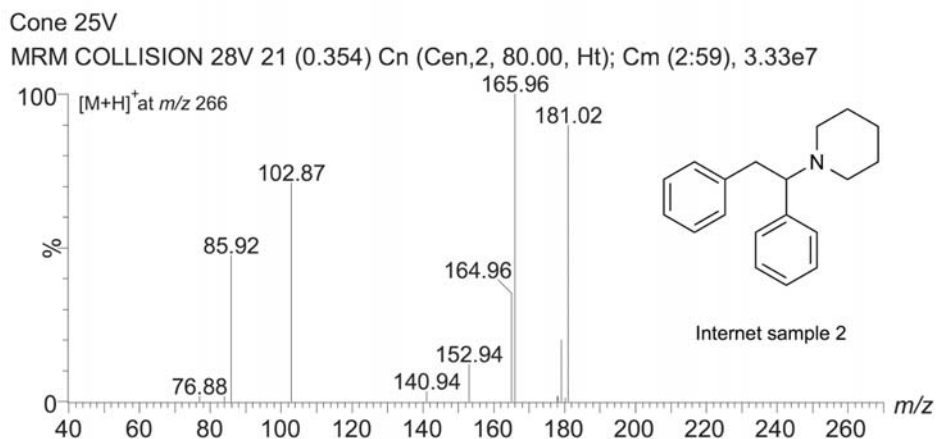
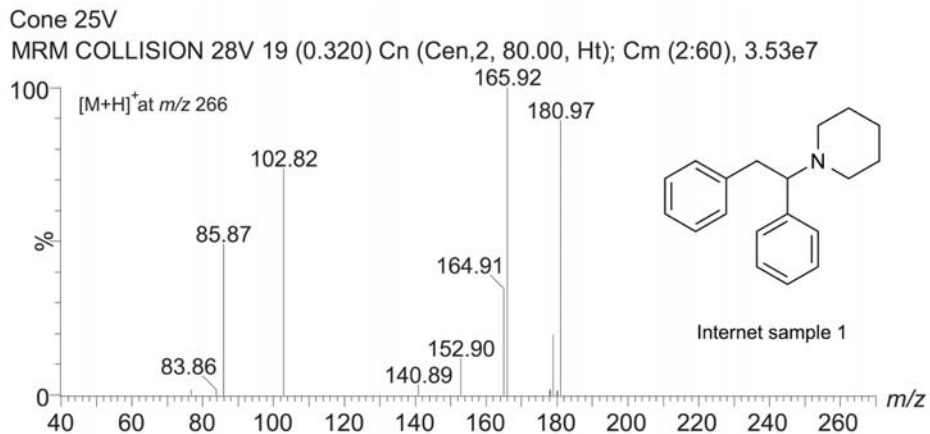
11) LC-ESI-MS/MS traces of synthesized **(1)** and **(2)** and two diphenidine samples obtained online

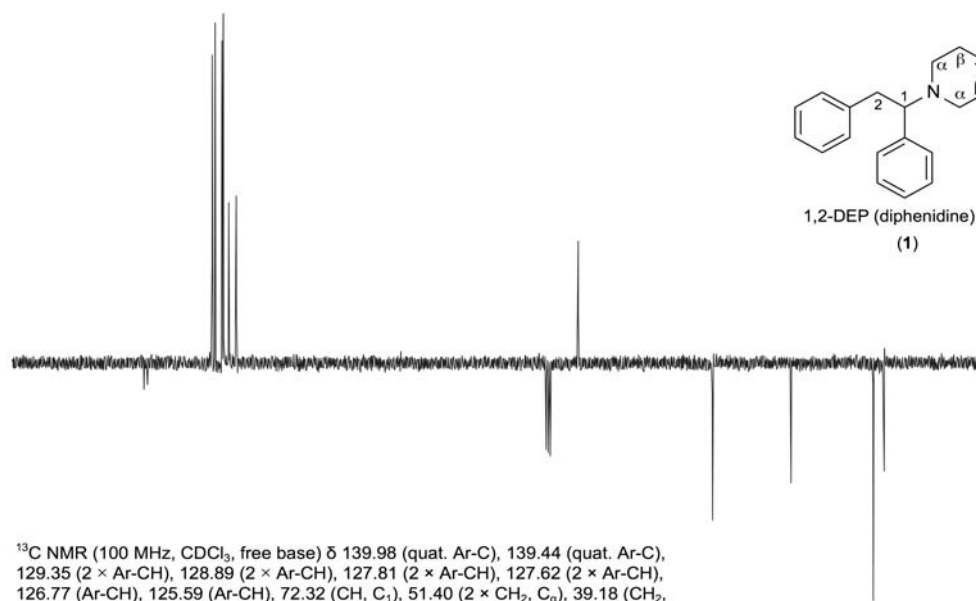


Selected ion transitions:

- $m/z$  266 > 72 (48 eV)
- $m/z$  266 > 103 (35 eV)
- $m/z$  266 > 166 (28 eV)
- $m/z$  266 > 181 (20 eV)

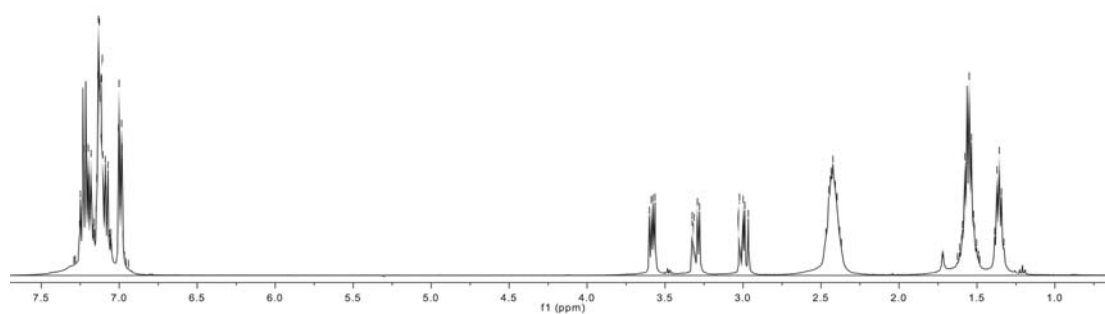
12) ESI-triple quadrupole tandem mass spectra of two diphenidine samples obtained online



13)  $^1\text{H}$  and  $^{13}\text{C}$  NMR of diphenidine free base

$^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ , free base)  $\delta$  139.98 (quat. Ar-C), 139.44 (quat. Ar-C), 129.35 (2  $\times$  Ar-CH), 128.89 (2  $\times$  Ar-CH), 127.81 (2  $\times$  Ar-CH), 127.62 (2  $\times$  Ar-CH), 126.77 (Ar-CH), 125.59 (Ar-CH), 72.32 (CH,  $\text{C}_1$ ), 51.40 (2  $\times$   $\text{CH}_2$ ,  $\text{C}_6$ ), 39.18 ( $\text{CH}_2$ ,  $\text{C}_2$ ), 26.37 (2  $\times$   $\text{CH}_2$ ,  $\text{C}_8$ ), 24.66 ( $\text{CH}_2$ ,  $\text{C}_7$ ).

$^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ , free base)  $\delta$  7.37–7.03 (8H, m, Ar-H), 7.04–6.95 (2H, m, Ar-H), 3.58 (1H, dd,  $J = 9.4, 5.2$  Hz,  $\text{C}_1\text{H}$ ), 3.30 (1H, dd,  $J = 13.3, 5.2$  Hz,  $\text{C}_2\text{H}$ ), 2.99 (1H, dd,  $J = 13.4, 9.4$  Hz,  $\text{C}_2\text{H}$ ), 2.55–2.29 (4H, m, 2  $\times$   $\text{C}_6\text{H}_2$ ), 1.64–1.45 (4H, m, 2  $\times$   $\text{C}_8\text{H}_2$ ), 1.36 (2H, quintet,  $J = 5.9$  Hz,  $\text{C}_7\text{H}_2$ ).



14) Assigned  $^{13}\text{C}$  and  $^1\text{H}$  aliphatic and chemical shifts in ppm for compounds (1) – (4) and (1) HCl.

$^{13}\text{C}$ Shift	1,2-DEP HCl (1)	1,2-DEP (1)	2,2-DEP (2)	1,2-DEPy (3)	2,2-DEPy (4)
C <sub>1</sub>	72.90	72.32	64.53	73.31	61.74
C <sub>2</sub>	36.75	39.18	48.89	42.96	50.87
C <sub>α</sub>	53.39 48.81	51.40	54.85	53.0	54.55
C <sub>β</sub>	22.71 22.65	26.37	25.99	23.35	23.50
C <sub>γ</sub>	22.24	24.66	24.43	-	-

Assigned  $^{13}\text{C}$  aliphatic chemical shifts in ppm for compounds (1) – (4) and (1) HCl.

$^1\text{H}$ Shift	1,2-DEP HCl (1)	1,2-DEP (1)	2,2-DEP (2)	1,2-DEPy (3)	2,2-DEPy (4)
C <sub>1</sub>	4.23 d (11.6, 1H)	3.58, dd (9.4, 5.2, 1H)	2.93, d (7.3, 2H)	3.30, dd (9.9, 4.3, 1H)	3.14, d (7.4, 2H)
C <sub>2</sub>	4.04 dd (12.8, 3.1, 1H)	3.30, dd (13.3, 5.2, 1H)	4.21, t (7.3, 1H)	3.36, dd (13.3, 4.3, 1H)	4.22, t (7.1, 1H)
C <sub>α</sub>	3.46 t (12.2, 1H) 3.64 d (10.3, 1H) 3.54 d (11.6, 1H) 2.68-2.39 m (2H)	2.99, dd (13.4, 9.4, 1H) 2.55-2.29, m (4H)	2.39, t (5.4, 4H)	2.96, dd (13.3, 9.9, 1H) 2.64, m (2H) 2.45, m (2H)	2.51, t (6.1, 4H)
C <sub>β</sub>	2.68-2.39 m (1H) 1.93-1.76 m (3H)	1.64-1.45, m (4H)	1.46, quint (5.5, 4H)	1.77, quint (3.3, 4H)	1.72, quint (3.1, 4H)
C <sub>γ</sub>	2.31 q (13.2, 12.2, 1H) 1.27 q (13.4, 12.4, 1H)	1.36, quint (5.9, 2H)	1.37, quint (5.4, 2H)	-	-

Assigned  $^1\text{H}$  aliphatic chemical shifts in ppm for compounds (1) – (4) and (1) HCl. m = multiplet; d = doublet; t = triplet; q = quartet; quint = quintet.



15) HPLC-DAD traces for synthesized diphenidine isomers (1) and (2) and comparison with a diphenidine sample obtained from an online test purchase. Note: The overlapping UV maxima for traces (a) and (c) were 257 and 267 nm, respectively.

Analytical details have been described in: Y.N. Soh, S. Elliott. An investigation of the stability of emerging new psychoactive substances. *Drug Test. Anal.* **2013**, in press; doi: 10.1002/dta.1576.

